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The Compound:

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What is  $K_{sp}$ ?

(Solubility Product

Constant)

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The benzyne intermediate forms when the bromide is expelled and the electrons on the  $sp^2$  orbital adjacent to it overlap with the empty  $sp^2$  orbital of the carbon that lost the bromide.

Benzynes are very reactive species due to the high strain of the triple bond. 59.

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Ethers, Epoxides, and

Sulfides 1) What is

the hybridization of

the oxygen atom in

dialkyl ethers? A)  $sp^3$

B)  $sp^2$  C)  $sp$  D)  $s$  E)  $p$

Answer: A Diff: 1

Section: 14.2 2)

Which of the

following

corresponds to the

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COC bond angle in dimethyl ether?

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1) Which of the following statements best describes the relative bond dissociation energies of the sigma and pi bonds ...

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