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question 1 If two

covalently bonded

atoms move closer

than a distance of the

bond length, the

potential energy of the

atoms a becomes

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negative b decreases

increases d remains
constant 2 The

electrons involved in

the formation of a

covalent bond are a

transferred from one

atom to another

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teaching requires
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educators have
struggled to teach this
discipline more

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provides

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science educators

with a path to

understanding

students,

accommodating their

individual differences,

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and helping them grasp the methods--and the wonder--of science.

What impact does teaching style have?

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questions and

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chemical bonding and

structure, chemical

formulae and

equations, electricity,

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Answers
chemicals, elements,
compounds, mixtures,

energy from

chemicals,

experimental

chemistry, methods of

purification, particles

of matter, redox

reactions, salts and

identification of ions

and gases, speed of

reaction, and

structure of atom to

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and Structure Multiple
Choice Questions: 75
MCQs Chemical
Formulae and
Equations Multiple
Choice Questions:
167 MCQs Electricity
Multiple Choice
Questions: 107 MCQs
Electricity and
Chemicals Multiple
Choice Questions: 10
MCQs Elements,

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Compounds and Test

Mixtures Multiple

Choice Questions: 39

MCQs Energy from

Chemicals Multiple

Choice Questions: 41

MCQs Experimental

Chemistry Multiple

Choice Questions: 18

MCQs Methods of

Purification Multiple

Choice Questions: 84

MCQs Particles of

Matter Multiple

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MCQs Redox

Reactions Multiple

Choice Questions: 42

MCQs Salts and

Identification of Ions

and Gases Multiple

Choice Questions: 61

MCQs Speed of

Reaction Multiple

Choice Questions: 35

MCQs Structure of

Atom Multiple Choice

Questions: 52 MCQs

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Chapter 6 "Acids
and Bases MCQs"
Answers

covers topics of acid
rain, acidity needs
water, acidity or
alkalinity, acids
properties and
reactions, amphoteric
oxides, basic acidic
neutral and
amphoteric, chemical
formulas, chemical
reactions, chemistry
reactions, college

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Chemistry, mineral
acids, general
properties,

neutralization,
ordinary level

chemistry, organic

acid, pH scale, acid

and alkali, properties,

bases and reactions,

strong and weak

acids, and universal

indicators. The

chapter "Chemical

Bonding and

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Structure MCQs"

covers topics of ions and ionic bonds, molecules and covalent bonds, evaporation, ionic and covalent substances, ionic compounds, crystal lattices, molecules and macromolecules, organic solvents, polarization, and transfer of electrons.

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"Chemical Formulae
and Equations MCQs"

covers topics of
chemical formulas,
chemical equations,
atomic mass, ionic
equations, chemical
reactions, chemical
symbols, mixtures
and compounds,
molar mass, percent
composition of
elements, reactants,

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relative molecular mass, valency and chemical formula, and valency table. The chapter "Electricity MCQs" covers topics of chemical to electrical energy, applications of electrolysis, reactions, conductors and non-conductors, dry cells, electrical devices, circuit symbols,

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Answers
electrolytes, non-electrolytes, organic solvents, polarization, and valence

electrons. The

chapter "Electricity

and Chemicals

MCQs" covers topics

of chemical to

electrical energy, dry

cells, electrolyte, non-

electrolyte, and

polarization. The

chapter "Elements,

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Compounds and Mixtures MCQs"

Answers
covers topics of

elements,

compounds, mixtures,

molecules, atoms,

and symbols for

elements.

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Weight and Formula

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Reactant Calculations

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Phase Equilibrium

Properties of Liquids

Density Colligative

Properties of

Solutions Raoult's

Law and Vapor

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Concentration Units

Equilibrium The Law
of Mass Action

Kinetics and

Equilibrium Le

Chatelier's Principle
and Chemical

Equilibrium Acid-Base

Equilibria Definitions

of Acids and Bases

Ionization of Water,

pH Dissociation of

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Weak Electrolytes
Dissociation of
Polyprotic Acids

Buffers Hydrolysis

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Bond Energies Some
Commonly Used

Terms in

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Bonds Organic

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Structure Alkanes

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Alkynes Alkyl Halides

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Epoxides Alcohols

and Glycols

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Carboxylic Acid

Derivatives Esters

Amides Arenes

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Ketones Amines

Phenols and

Quinones Structural

Isomerism SIX

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SAT II: Chemistry is a one-hour exam

consisting of 85

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multiple-choice questions. The first part of the exam

consists of

classification

questions. This

question type

presents a list of

statements or

questions that you

must match up with a

group of choices

lettered (A) through

(E). Each choice may

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Answers

be used once, more than once, or not at all. The exam then shifts to relationship analysis questions which you will answer in a specially numbered section of your answer sheet.

You will have to determine if each of two statements is true or false and if the second statement is a

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correct explanation of
the first. The last
section is composed

strictly of multiple-
choice questions with
choices lettered (A)

through (E). Material

Tested The following
chart summarizes the

distribution of topics

covered on the SAT

II: Chemistry Subject

Test. Topic /

Percentage / Number

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of Questions Atomic &

Molecular Structure /

25% / 21 questions

States of Matter / 15%

/ 13 questions

Reaction Types / 14%

/ 12 questions

Stoichiometry / 12% /

10 questions

Equilibrium &

Reaction Times / 7% /

6 questions

Thermodynamics /

6% / 5 questions

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Descriptive Chemistry

/ 13% / 11 questions

Laboratory / 8% / 7

questions The

questions on the SAT

II: Chemistry are also

grouped into three

larger categories

according to how they

test your

understanding of the

subject material.

Category / Definition /

Approximate

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Chapter 6 Test 1)

Factual Recall /

Demonstrating a

knowledge and

understanding of

important concepts

and specific

information / 20% 2)

Application / Taking a

specific principle and

applying it to a

practical situation /

45% 3) Integration /

Inferring information

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and drawing
conclusions from
particular

relationships / 35%

STUDYING FOR THE

SAT II: CHEMISTRY

It is very important to
choose the time and
place for studying that
works best for you.

Some students may
set aside a certain
number of hours
every morning to

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study, while others may choose to study at night before going to sleep. Other

students may study

during the day, while

waiting on line, or

even while eating

lunch. Only you can

determine when and

where your study time

will be most effective.

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your time wisely.

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Answers

test-taking anxieties.

Become comfortable with the format of the exam. When you are practicing to take the SAT II: Chemistry Subject Test, simulate the conditions under which you will be taking the actual test. Stay calm and pace yourself. After simulating the test only a couple of

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Chapter 6 Test
Answers

times, you will boost your chances of doing well, and you will be able to sit down for the actual exam with much more confidence. Know the directions and format for each section of the test. Familiarizing yourself with the directions and format of the exam will not only save you time,

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but will also ensure
that you are familiar
enough with the SAT

II: Chemistry Subject
Test to avoid

nervousness (and the
mistakes caused by

being nervous). Do

your scratchwork in

the margins of the test

booklet. You will not

be given scrap paper

during the exam, and

you may not perform

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Scratchwork on your answer sheet. Space is provided in your test booklet to do any necessary work or draw diagrams. If you are unsure of an answer, guess.

However, if you do guess - guess wisely. Use the process of elimination by going through each answer to a question and

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ruling out as many of the answer choices as possible. By eliminating three answer choices, you give yourself a fifty-fifty chance of answering correctly since there will only be two choices left from which to make your guess. Mark your answers in the appropriate spaces on

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the answer sheet. Fill in the oval that corresponds to your answer darkly, completely, and neatly. You can change your answer, but remember to completely erase your old answer. Any stray lines or unnecessary marks may cause the machine to score your answer incorrectly.

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When you have finished working on a section, you may want to go back and check to make sure your answers correspond to the correct questions. Marking one answer in the wrong space will throw off the rest of your test, whether it is graded by machine or by hand. You don't

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have to answer every question. You are not penalized if you do not answer every question. The only penalty results from answering a question incorrectly. Try to use the guessing strategy, but if you are truly stumped by a question, remember that you do not have to answer it. Work

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quickly and steadily.

You have a limited amount of time to work on each section, so you need to work quickly and steadily. Avoid focusing on one problem for too long. Before the Test Make sure you know where your test center is well in advance of your test day so you do not get lost on the day of

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Chapter 10 Test
Answers

the test. On the night before the test, gather together the materials you will need the next day: - Your admission ticket - Two forms of identification (e.g., driver's license, student identification card, or current alien registration card) - Two No. 2 pencils with erasers - Directions to the test

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Chapter - A watch (if

you wish) but not one
that makes noise, as

it may disturb other

test-takers On the day

of the test, you should

wake up early (after a

good night's rest) and

have breakfast. Dress

comfortably, so that

you are not distracted

by being too hot or

too cold while taking

the test. Also, plan to

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Chapter 6 Test

Answers

arrive at the test center early. This will allow you to collect

your thoughts and relax before the test,

and will also spare you the stress of

being late. If you

arrive after the test begins, you will not be

admitted to the test center and you will

not receive a refund.

During the Test When

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you arrive at the test center, try to find a seat where you feel most comfortable.

Follow all the rules and instructions given by the test supervisor. If you do not, you risk being dismissed from the test and having your scores canceled. Once all the test materials are passed out, the test instructor

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Answers
will give you
directions for filling
out your answer

sheet. Fill this sheet
out carefully since this
information will

appear on your score
report. After the Test

When you have

completed the SAT II:

Chemistry Subject

Test, you may hand in

your test materials

and leave. Then, go

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Chapter 9 Test

Answers

home and relax!
When Will I Receive My Score Report and What Will It Look Like? You should receive your score report about five weeks after you take the test. This report will include your scores, percentile ranks, and interpretive information.

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plays in their daily

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